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## **Calculate the heat required to melt 25.7 g of solid ...**

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### **A 155-g sample of an unknown substance was heated from ...**

Chapter 01 = Introduction to Chemistry



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Figure 15.4.1 Energy Changes Accompanying the Thermite Reaction Because enthalpy is a state function, the overall enthalpy change for the reaction of 2 mol of Al(s) with 1 mol of Fe<sub>2</sub>O<sub>3</sub>(s) is  $-851.1$  kJ, whether the reaction occurs in a single step ( $\Delta H$  4, shown on the left) or in three hypothetical steps (shown on the right) that involve ...

## **Chapter 15.4: Hess's Law - Chemistry LibreTexts**

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